



Stronger Greenhouse Effect on Early Mars: Collision-Induced Absorption by CO₂-H₂ and CO₂-CH₄ Complexes

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Canadian
Light
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Introduction

- Geological evidence on Mars strongly suggests there was once liquid water on its surface.
- Most atmospheric models of ancient Mars have difficulty producing sufficient warming to raise the surface temperature above 0°C.
- A recent study by Wordsworth *et al.* [1] suggested that previously unaccounted-for collision-induced absorption (CIA) by carbon dioxide (CO₂) and hydrogen gas (H₂), and by CO₂ and methane (CH₄) could provide the additional atmospheric absorption needed to trap enough radiation to raise the Martian surface temperature above freezing.
- CIA cross-sections for CO₂-H₂ and CO₂-CH₄ complexes do not exist in the literature over the full spectral and temperature range desired, so the authors could only use computational methods to simulate the CIA absorption cross-sections that they themselves identify in the study as needing experimental validation.
- Recent work done by Turbet *et al.* [2], have investigated CIA of CO₂-H₂ and CO₂-CH₄ complexes at room temperature in the range of 40-640 cm⁻¹.

Theory

- Collisions between molecules momentarily form a two-molecule complex that has its own absorption spectrum; this is known as collision induced absorption.

- CIA are typically broad band features.

- The absorption of light passing through a gas mixture is given by Beer's Law:

$$I = I_0 e^{-L(\rho_{CO_2} \sigma_{CO_2} + \rho_x \sigma_x + \rho_{CO_2}^2 \sigma_{CO_2+CO_2} + \rho_x^2 \sigma_{x+x} + \rho_{CO_2} \rho_x \sigma_{CO_2+x})}$$

where I_0 is the intensity of light before passing through the mixture, I is the intensity of light after absorption, L is the pathlength of light through the mixture, ρ_{CO_2} is the density of CO₂, ρ_x is the density of either H₂ or CH₄, and the σ are the absorption cross-sections for individual molecule absorption or CIA absorption depending on the subscript.

- Single species cross-sections are known, as are some of the self CIA cross-sections.
- By filling the gas cell with a single gas first, I_0 can absorb the terms of the exponential that only depend on that gas.

Experimental Details

- Experiments were performed at the FAR-IR beamline of the Canadian Light Source (CLS) Synchrotron.
- Applications for beamtime at the CLS is a highly competitive process; this project was awarded two weeks of beamtime.
- The FAR-IR beamline has a White cell that was used to detect the weak CIA absorption features.
- The White cell has a maximum pathlength of 7274.93±6 cm, and a maximum pressure of 1 atm.

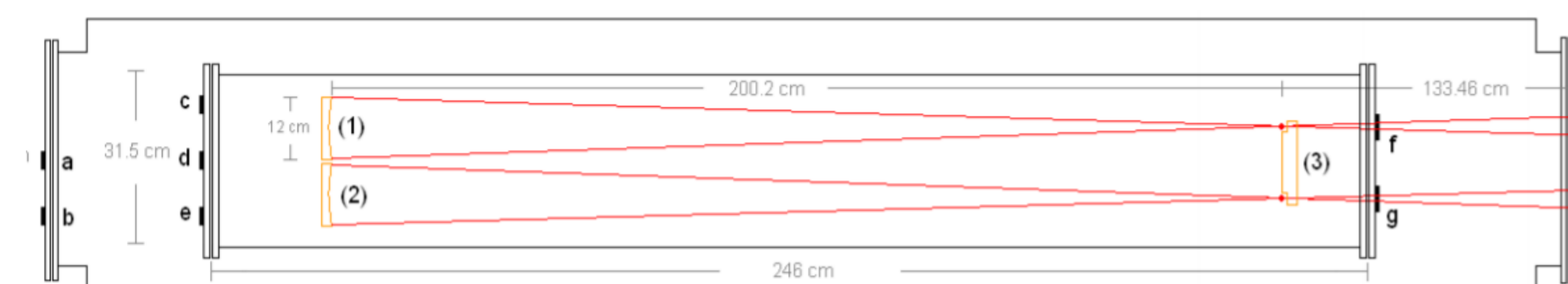


Fig. 1: White cell schematic. Mirrors are placed at both ends of the cell. A beam of light bounces between the mirror multiple times before exiting the cell, allowing for a pathlength much longer than the length of the cell. The cell is nested inside a second vacuum chamber to isolate it from environmental affects.

- Safety protocols at the CLS limit explosive gas mixtures to maximum 20% CH₄ and 8.3% H₂.
- No combination of beamsplitter, windows, and detector covers the full spectral range, so the experiment is divided into two optical regimes:
 - Below 600 cm⁻¹, polypropylene windows, Si bolometer detector, and mylar beamsplitter are used.
 - Above 500 cm⁻¹, KBr windows and beamsplitter, and MCT detector are used.

Carbon Dioxide CIA

- Experimental measurements of CO₂-CO₂ CIA at three different temperatures.
- Strange signal observed around 70 cm⁻¹, unclear what caused it.
- General agreement with calculations by Gruszka and Borysow [3].
- 250 K result weaker than expected.

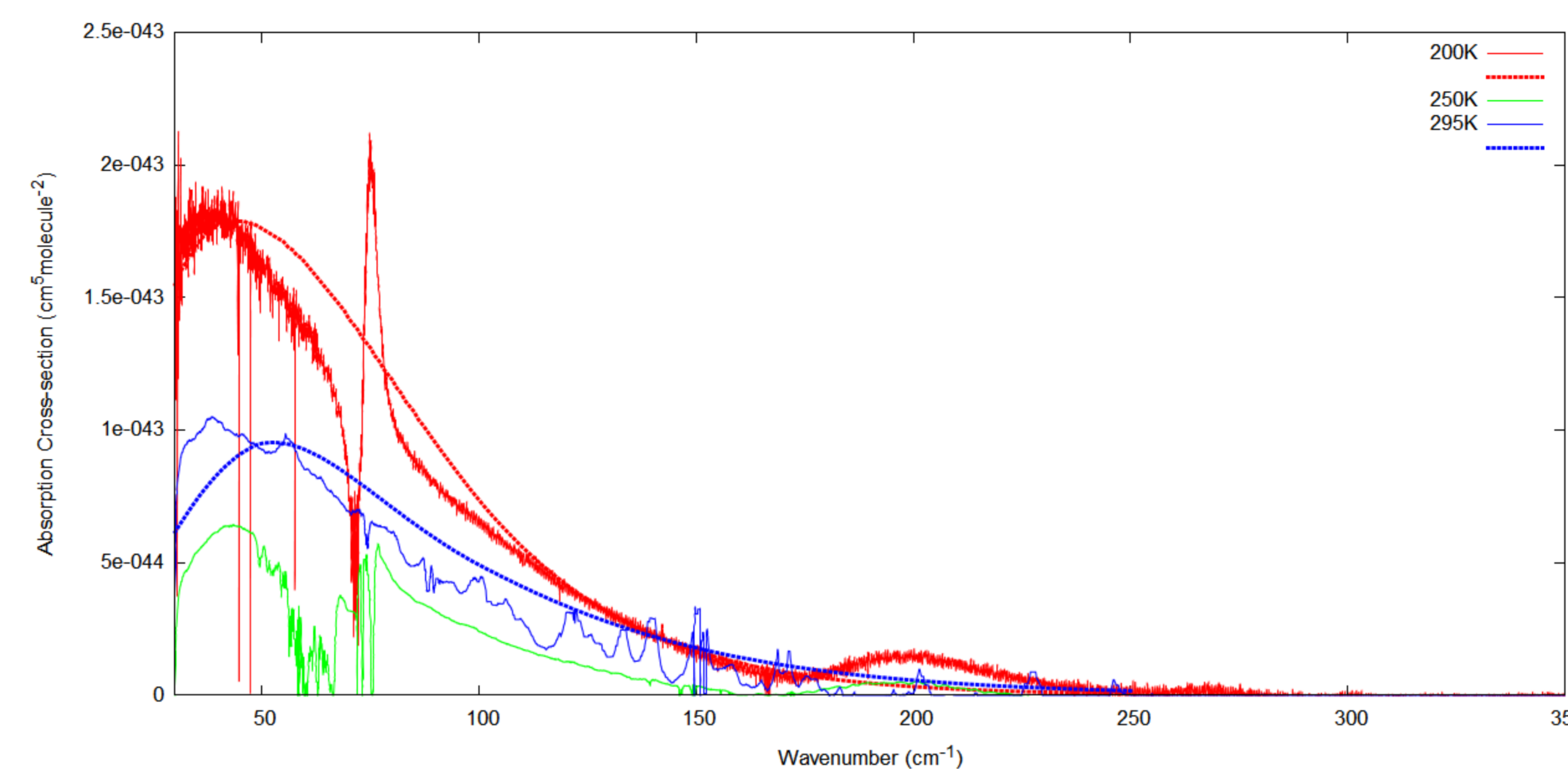


Fig. 2: CO₂-CO₂ CIA from this work (solid lines) along with comparisons to Gruszka and Borysow (dotted lines) [3].

Methane-Carbon Dioxide CIA

- Experimental measurements of CO₂-CH₄ CIA at three different temperatures.
- 250 K result stronger than expected, although this is likely due to insufficient removal of CO₂-CO₂ CIA due to the weak measurement found in the above section.
- Agreement with Turbet *et al.* [2] that the experimental results are weaker than the theoretical predictions of Wordsworth *et al.* [1].
- Difficulty removing the single molecule absorption by CH₄ and CO₂ is causing anomalous intensity values above 600 cm⁻¹.

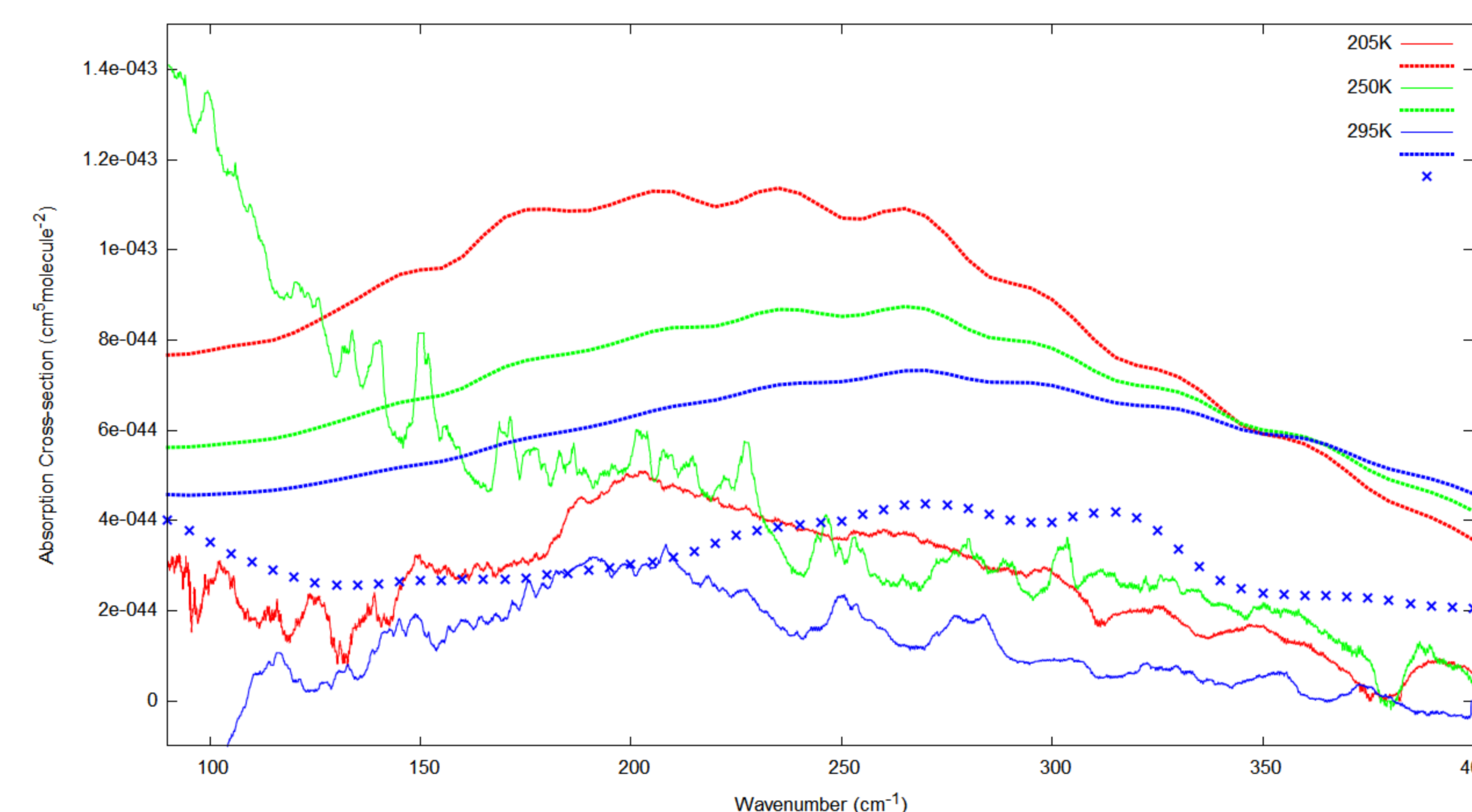


Fig. 3: CO₂-CH₄ CIA from this work (solid lines) along with comparisons to Wordsworth *et al.* (dotted lines) [1] and Turbet *et al.* (X line) [3].

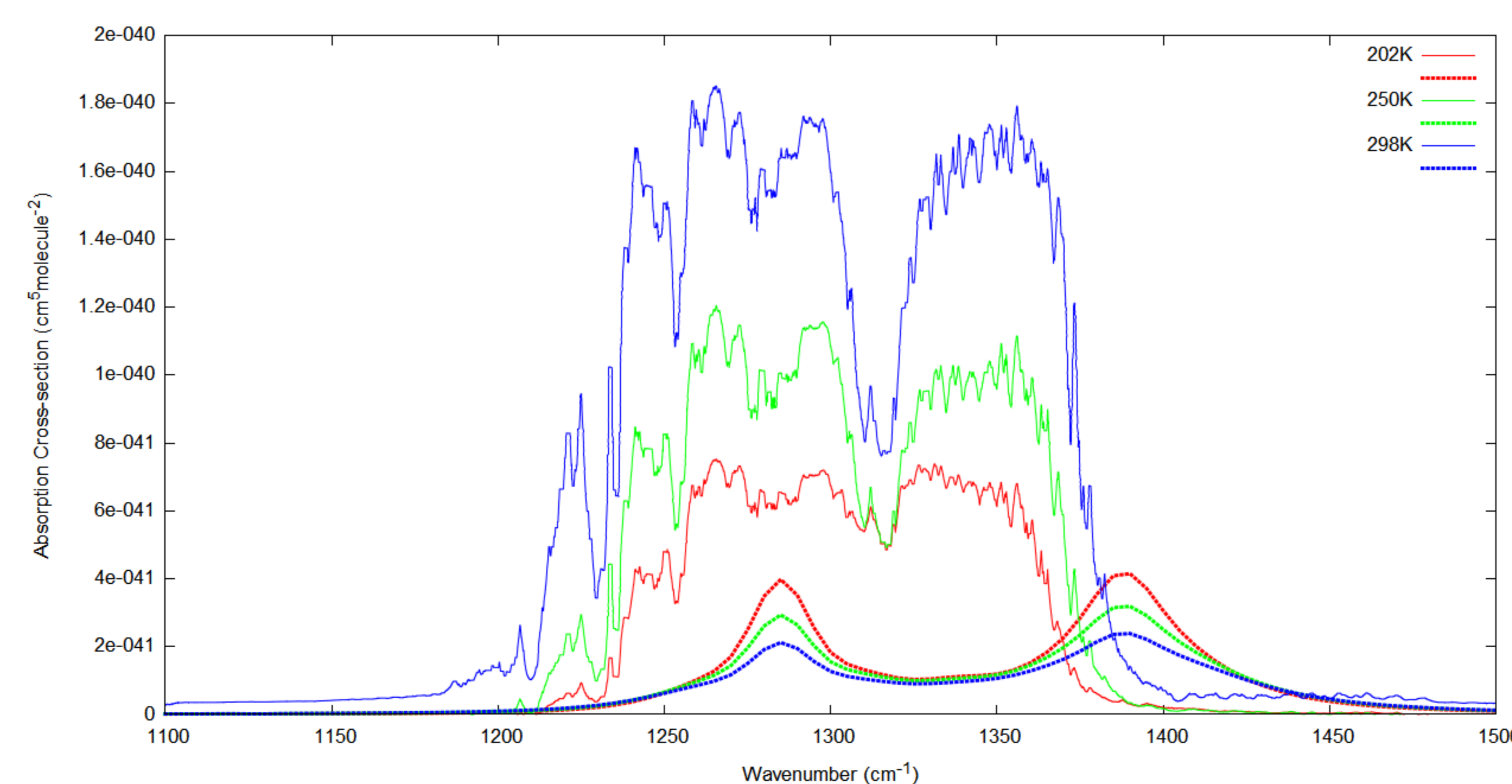


Fig. 4: CO₂-CH₄ CIA from with work (solid lines) along with comparisons to Wordsworth *et al.* (dotted lines) [1], which are increased by a factor of 100. CO₂-CO₂ CIA is assumed to be 0 in this spectral range.

Hydrogen-Carbon Dioxide CIA

- Experimental measurements of CO₂-H₂ CIA at three different temperatures.
- Significant water contamination present in the 250 K measurement.
- Agreement with Turbet *et al.* [2] that the experimental results are weaker than the theoretical predictions of Wordsworth *et al.* [1] below 600 cm⁻¹.
- Absorption by CO₂ is causing anomalous intensity values from 550 to 1000 cm⁻¹.

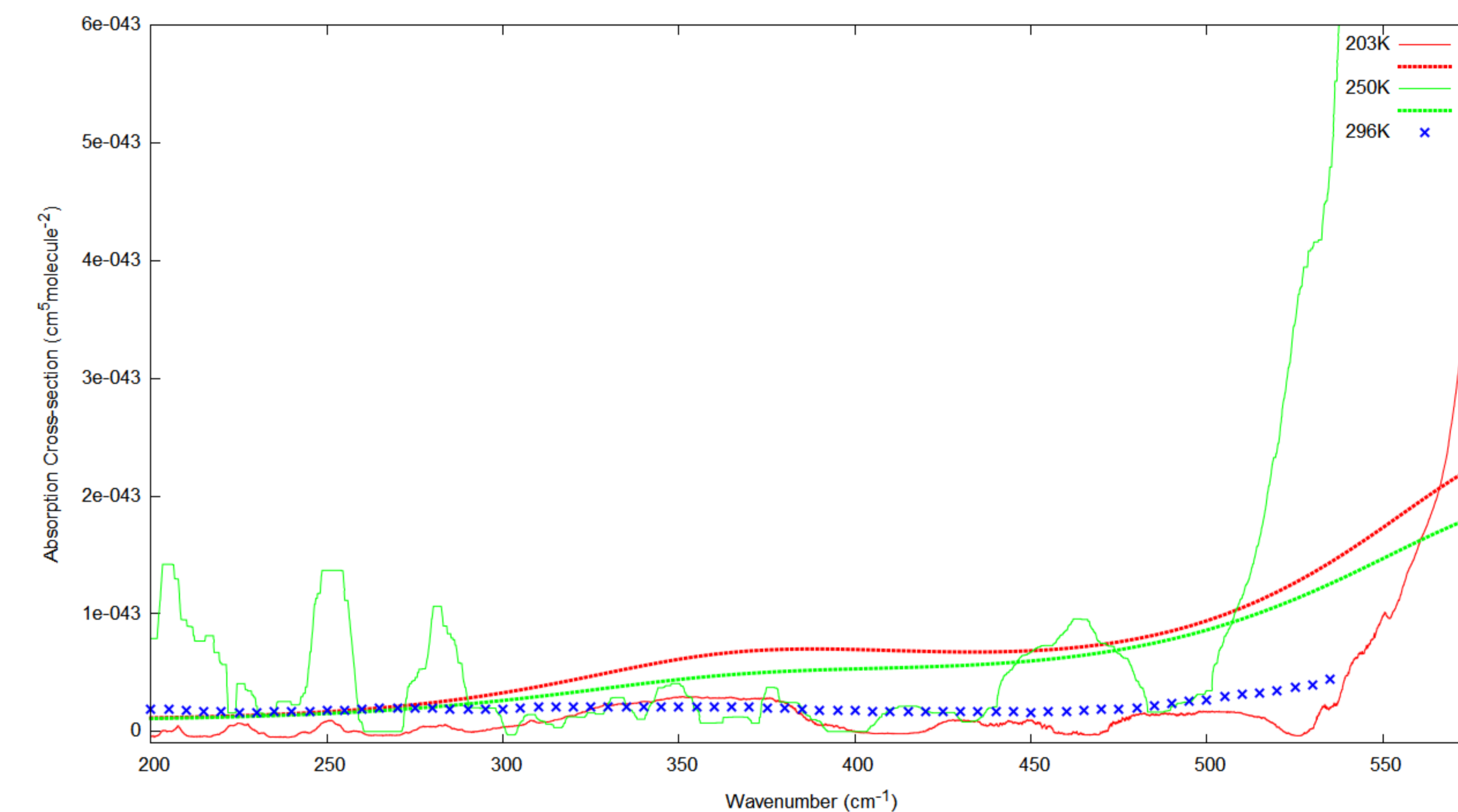


Fig. 5: CO₂-H₂ CIA from this work (solid lines) along with comparison to Wordsworth *et al.* (dotted lines) [1] and Turbet *et al.* (X line) [3].

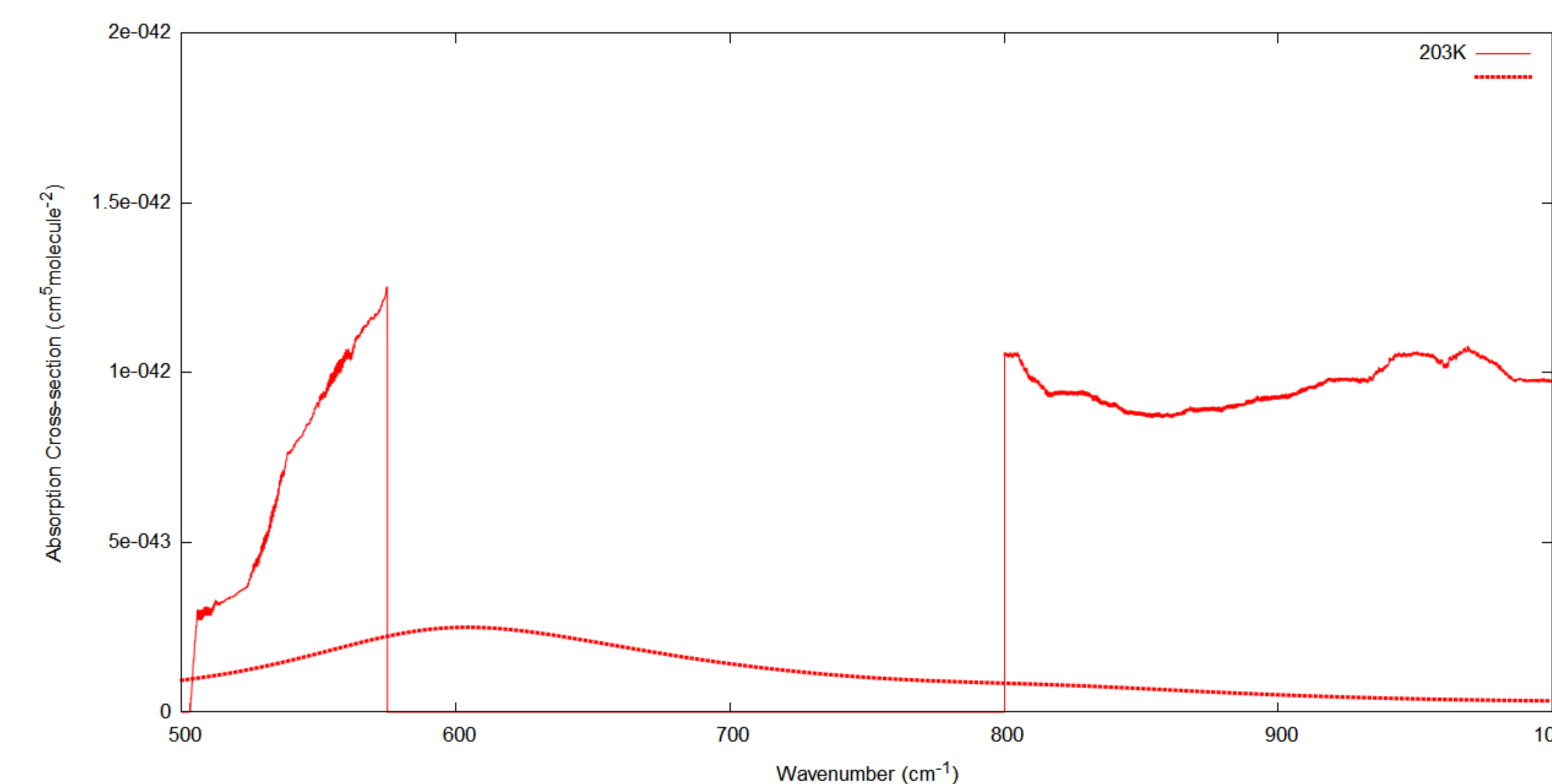


Fig. 6: CO₂-H₂ CIA from this work (solid line) along with comparison to Wordsworth *et al.* (dotted line) [1]. Due to complete absorption by CO₂, nothing can be resolved in the range of 575-800 cm⁻¹. CO₂-CO₂ CIA is assumed to be 0 in this spectra range.

Conclusions

- First temperature dependent experimental measurements of CO₂-H₂ and CO₂-CH₄ CIA.
- First experimental measurements of CO₂-H₂ and CO₂-CH₄ CIA above 600 cm⁻¹.
- CO₂-CO₂ CIA results agree with the calculations by Gruszka and Borysow [3] at room temperature. However at lower temperatures small scale structure is seen, although overall shape and amplitude agrees with their theoretical calculations.
- Below 600 cm⁻¹, CO₂-H₂ and CO₂-CH₄ CIA agrees with the position and general shape of the calculation by Wordsworth *et al.* [1], however the amplitude is weaker than predicted.
- Above 600 cm⁻¹, CO₂-CH₄ CIA does exhibit the two peak structure as predicted, but it is shifted to lower wavenumbers. Due to difficulty removing the single molecule absorption by CH₄ and CO₂, constraining the amplitude of the CIA absorption is on going.
- Above 600 cm⁻¹, CO₂-H₂ CIA does appear to exhibit the CIA absorption structure as predicted, but due to saturation by CO₂ the full nature of the absorption remains unknown.