

Trapping the Short-Chain Odd Carbon Number Olefins Using Nickel(II)-catalyzed Tandem Ethylene Oligomerization and Friedel-Crafts Alkylation of Toluene

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Abstract

Nickel(II) complexes with pyrazole-based ligands are widely employed in catalysis of ethylene oligomerization and subsequent Friedel-Crafts alkylation of toluene. We have prepared ten new nickel(II) dibromide complexes with various substituted bis(azolyl)methanes. They have been characterized using ¹H NMR, IR, high resolution mass spectrometry and elemental analysis. The structures of three complexes have been unambiguously established using X-ray diffraction. It was found that these complexes in the presence of Et₂AlCl or Et₃Al₂Cl₃ are active both in ethylene oligomerization and Friedel-Crafts alkylation processes (activity up to 3720 kgoligomer·mol[Ni]·¹h⁻¹). The use of Et₃Al₂Cl₃ results in the higher share of alkylated products (up to 60%). Moreover, catalytic systems activated with Et₃Al₂Cl₃ produced small amounts of odd carbon number olefins (up to 0.8%). The Friedel-Crafts alkylation was used as a trap for previously undetected short-chain odd carbon number olefins (C₃ and C₅).

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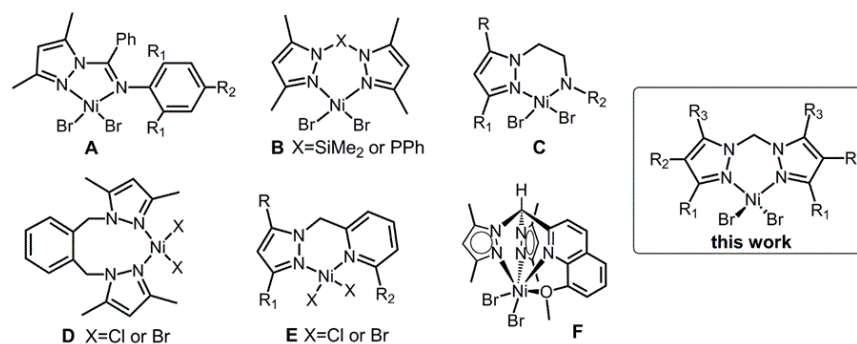
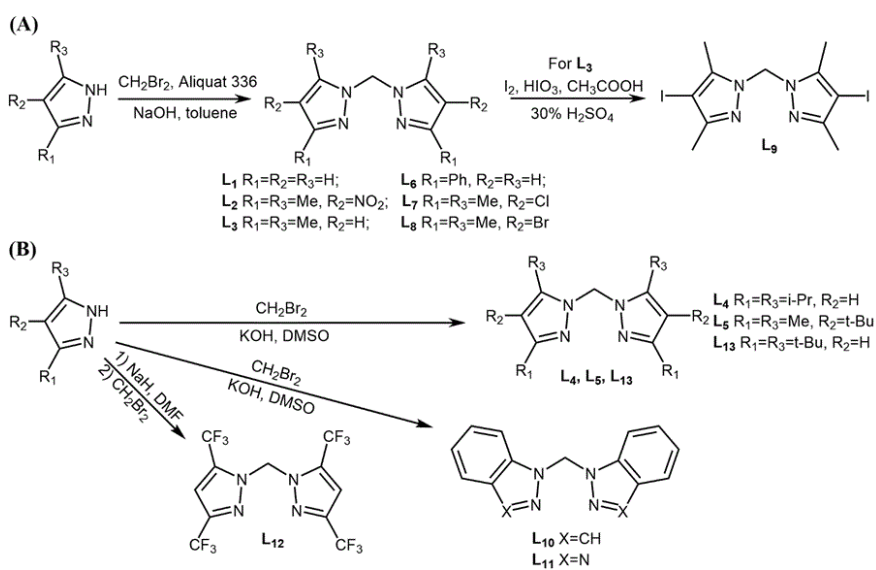
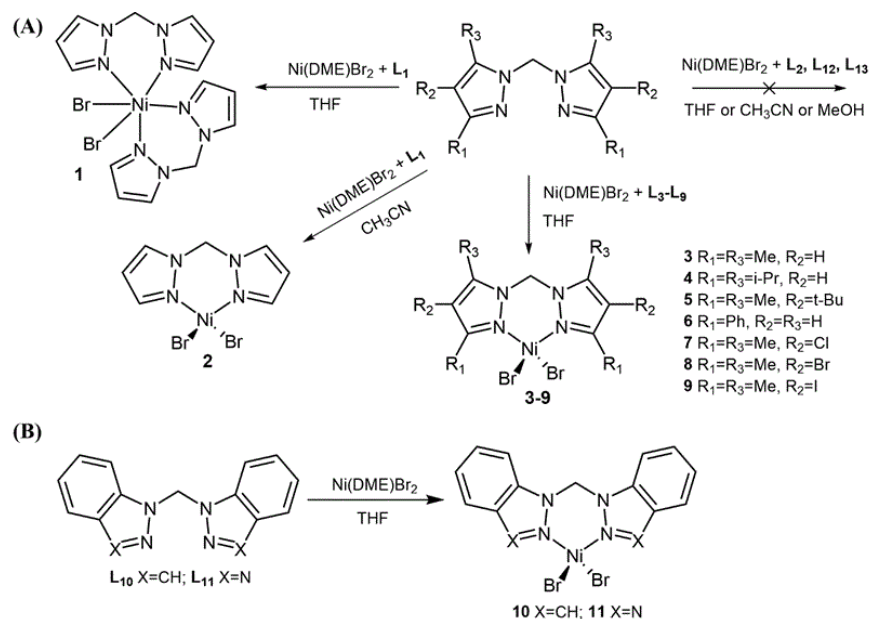


Figure 1. Structures of nickel(II)-halide complexes with pyrazole-containing ligands active in ethylene oligomerization.¶



Scheme 1. Synthesis of bis(azolyl)methane ligands L_1 - L_{13} .¶



Scheme 2. Synthesis of nickel(II) complexes 1-11.

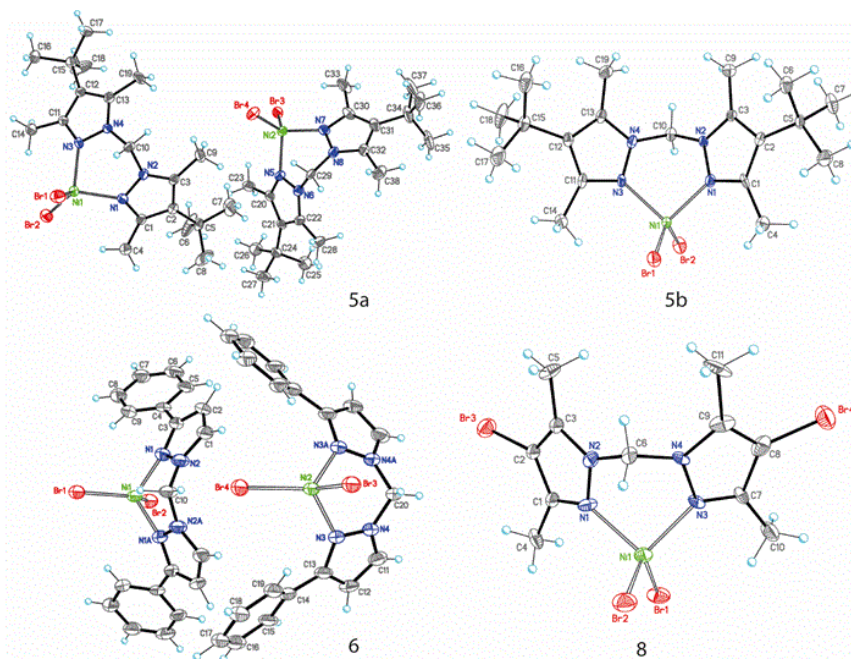


Figure 2. Molecular structures of complexes 5a, 5b, 6 and 8. Thermal ellipsoids are shown at the 50% probability level.

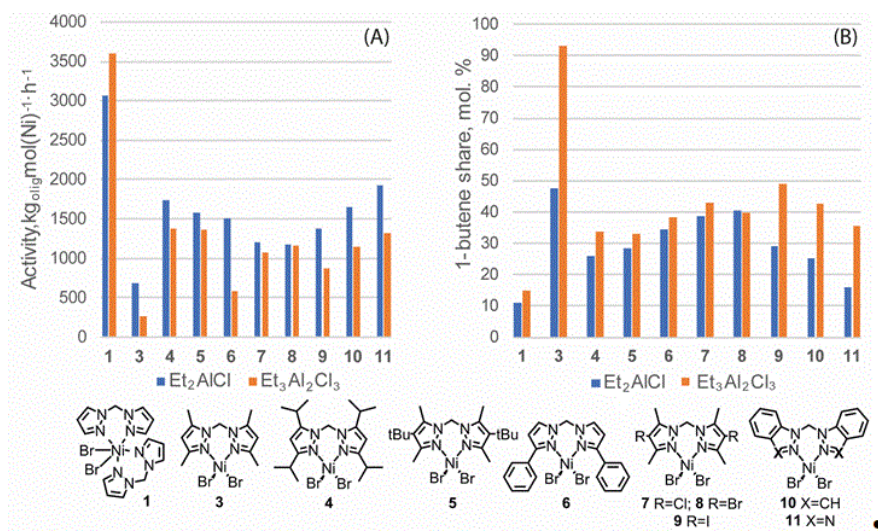


Figure 3. Dependencies of catalytic activity (A) and 1-butene share (B) on the composition of catalytic system. ¶

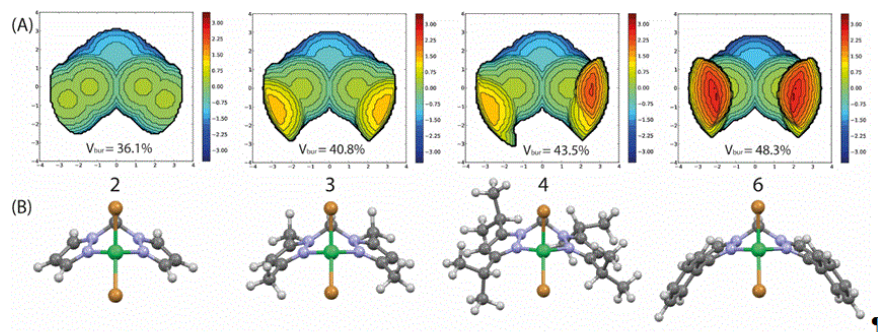


Figure 4. (A): Topographic steric maps of the nickel complexes 2, 3, 4 and 6. Only bispyrazolylmethane ligands have been considered for calculations of % V_{bur} . Ni atom was placed as the center of the sphere and xz-plane was defined by Ni and 2 coordinated nitrogen atoms; the z-axis bisects N-Ni-N angle. (B): Structures of nickel complexes 2, 3, 4 and 6 optimized using DFT calculation and oriented the same way as topographic steric maps. ¶

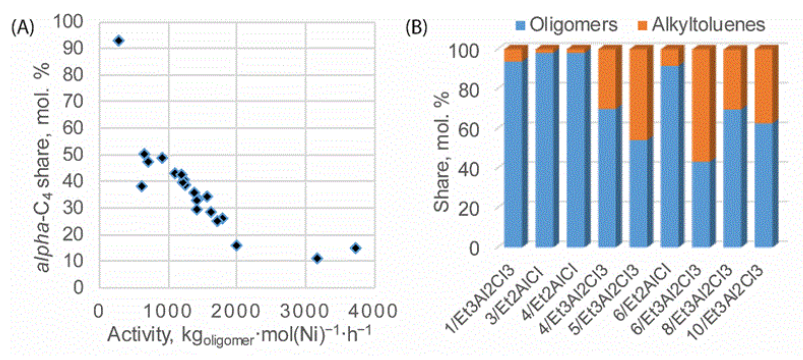


Figure 5. (A) Dependence of 1-butene share on catalytic activity; (B) shares of oligomers and alkyltoluenes for catalytic systems based on complexes 1, 3, 6, 8, 10 (determined using GC-MS). ¶

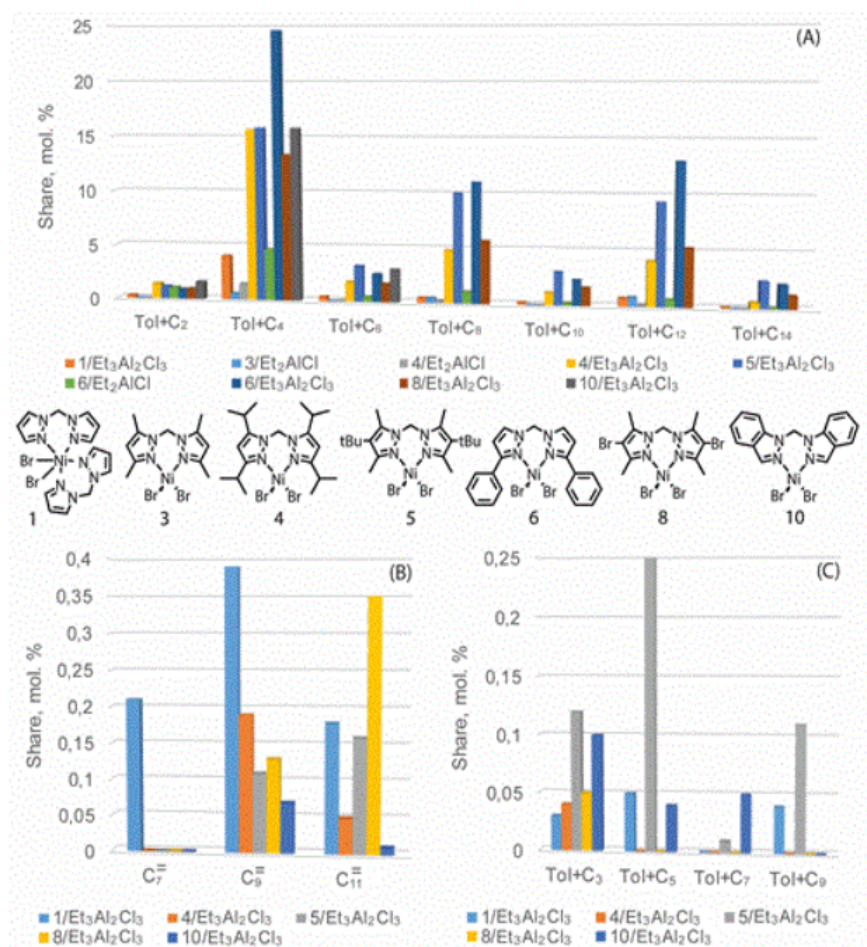


Figure 6. Shares of alkyltoluenes (A), odd carbon number olefins (B) and toluene alkylated with different odd carbon number olefins (C) in the reaction mixtures produced with catalytic systems based on complexes 1, 3, 6, 8, 10 (determined using GC-MS).¶